# A Mathematical Model of Middle Ear Ventilation. Controlled Valvular Tympanostomy Tube 

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#### Abstract

BAKGROUND The middle ear (tympanic cavity and mastoid process) is a closed cavity. It is filled with a special gas mixture and at the same pressure as atmospheric air, the composition of the gas is similar to the surrounding tissues. Its change in pressure is compensated by the occasional opening of the Eustachian tube. If the tube does not open, a persistently high negative pressure develops in the middle ear, the clinical consequences of which are well known. The negative pressure is eliminated by a tympanostomy tube implanted in the tympanic membrane. Inserting the tube instead of one abnormal condition (permanently closed middle ear) creates another abnormal condition (permanently open middle ear). Prolonged insertion of the tube may cause irreversible damage. / eardrum atrophy, tympanosclerosis, permanent perforation, auditory bone damage, cholesteatoma. AIM We are looking for a solution that is similar to the original, i.e. the middle ear should be closed and open from time to time. METHOD We studied middle ear ventilation based on literature data, our own research, and our computer model. RESULTS We concluded that the original function can be restored with the valve tube. In our case, the opening and closing of the valve is controlled by a change in volume caused /diethyl ether/ by a change in body temperature. CONCLUSION A more advanced solution would be a nanotechnology valve controlled by a change in total pressure, a partial pressure change in oxygen or carbon dioxide. However, a valve that can be opened at different intervals or even individually adjusted is conceivable.


Key words: middle ear ventilation; mathematical model; gas diffusion; valvular tympanostomy tube; tympanostomy tube

## Introduction

The middle ear /ME/ is a closed, special gas mixture filled cavity.[1-6] Closing is passive. The pressure changes of the ME are compensated by the actively opening Eustachian tube /ET/.
Otitis media is a common disease in childhood. Permanently decreased in middle ear pressure caused by ET dysfunction is one of the fundamental factors in the pathology of otitis media.

The ME / tympanic cavity and mastoid process / pressure is provided by gas diffusion and ET function. Gas diffusion between the surrounding tissues /ST/ and the ME takes place in accordance with the laws of physics (bidirectional gas exchange), so the pressure from the diffusion in the ME, can be a maximum of 706 mmHg . (Table 1) That is, $92 \%$ of the ME pressure is provided by the gas diffusion, and $8 \%$ by the ET (Table 2 ).

| GAS | MIDDLE EAR* | SURROUNDING TISSUES |
| :---: | :---: | :---: |
| $\mathrm{O}_{2}$ | 51 | 40 |
| $\mathrm{~N}_{2}$ | 616 | 573 |
| $\mathrm{CO}_{2}$ | 46 | 46 |
| TOTAL | 713 | $/-54 /$ |

The data refer to dry gases
*Optimal gas composition in the middle ear

## Table 1: Pressure from Gas Diffusion in the Middle Ear

| Partial pressure of gases |  |  |  |
| :---: | :---: | :---: | :---: |
| /Hgmm/ |  |  |  |
| Gas | Air | Middle ear | Surrounding tissues |
| $\mathrm{O}_{2}$ | 158 | 51 | 40 |
| $\mathrm{CO}_{2}$ | 0,3 | 46 | 46 |
| $\mathrm{N}_{2}$ | 596 | 616 | 573 |
| Vapor | 5,7 | 47 | 47 |
| Total | 760 | 760 | 706 |

Table 2: In normal operation, the pressure of the ME gas is the same as the atmospheric pressure, and the gas composition is similar to the gas composition of the ST (3-5 Table).

## COMPOSITION OF MIDDLE EAR GASES

| GAS | PRESSURE mmHg | VOLUME ml | VOLUME \% |
| :--- | :---: | :---: | :---: |
| Oxigen | 51 | 405 | 7,10 |
| Carbon dioxid | 46 | 363 | 6,45 |
| Nitrogen | 616 | 4861 | 86,34 |
| Total | 713 | 5629 | 99,89 |

The data refer to dry gases and a 6000 ml middle ear.

## Table 3

## GAS COMPOSITION OF MIDDLE EAR

| GAS | O2 | CO2 | N2 |
| :--- | :---: | :---: | :---: |
| CALCULATED | 51 | 46 | 616 |
| MEASURED* | $39-54$ | $48-52$ | $563-606$ |
| *according to several authors |  |  |  |

Table 4
/ Partial pressure of gases in mmHg with respect to dry gases /

| Gas | Air | Middle ear* |  | Surroundin |  |
| :--- | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{O}_{2}$ | 150 | $\Rightarrow 138$ | $\rightarrow$ | 40 |  |
| $\mathrm{CO}_{2}$ | - | $\Leftarrow 15$ | $\leftarrow$ | 46 |  |
| $\mathrm{~N}_{2}$ | 563 | $\Leftarrow 570$ | $\leftarrow$ | 573 |  |
| Total | 713 | 713 | $/-54 /$ | 659 |  |

$\Rightarrow$ diffusion acros the tympanostomy tube
$\rightarrow$ diffusion through the mucosa

* FELDING és mtsai. /3/


## Table 5: GAS DIFFUSION AFTER TUBE INSERTION

The pressure and gas composition are provided by complex active and passive mechanisms operating in harmony with each other, including the nasopharyngeal biphasic pressure wave, the mastoid process volume and the tension and elasticity of the eardrum. ME volume is an important pathological factor. Diffusion of the same quantity of gas ( $\mathrm{O}_{2}, \mathrm{~N}_{2}, \mathrm{CO}_{2}$ ) causes a greater pressure change in a smaller volume middle ear.
If the ET does not open, the partial pressures of the gases / $\mathrm{O}_{2}, \mathrm{~N}_{2} /$ are equalized between the ME and ST and negative pressure develops in the middle ear with well-known clinical consequences.
The importance of negative ME pressure in pathology is demonstrated by many millions of measurements / tympanometry / daily around the world. The pathological process attempts are made to prevent by inserting a tympanostomy tube.
The long-term effects of tube insertion have been analyzed in several studies [7-10].

## The consequences of tube insertion

After tube insertion, the pressure in the ME is restored but the gas composition of the ME changes [11-13]. The gas composition of the ME will be more similar to air than to the gas composition of the ST. The larger lumen of the tube, the more the gas composition of ME resembles air rather than the gas composition of ST (Table 5).
Intermittent pressure equalization, which under intact conditions is the result of a dynamic process and intermittent, will be continuous and passive. This passivity can lead to structural damage.
The partial pressure of oxygen in the ME is too high, therefore the absorption of oxygen into the ST increases / hyperoxia /.

Strong, continuous diffusion of carbon dioxide begins from the surrounding tissues into the cavity of the ME, causing the ST to lose large amounts of $\mathrm{CO}_{2}$.
The ME loses water vapor / $100 \%$ humidity in the intact middle ear / thus the mucous membrane dries out and is damaged.
Inserting the ventilation tube instead of one abnormal condition (permanently closed middle ear), does another (permanently open middle ear).
So we did not restore the original condition by inserting the tube. The ear may compensate for our therapy in the short term, but with repeated or prolonged tube insertion, side effects may occur (atrophy, scarring, tympanosclerosis, perforation of the eardrum, damage to the auditory ossicles, cholesteatoma, and hearing loss).

## AIMS

We are looking for a solution that creates conditions similar to intact ventilation, i.e. the middle ear should be closed and open from time to time. We examined the ventilation of the middle ear based on known literature data and our own research. Intact middle ear ventilation was analyzed using a mathematical model.

## METODS

Our calculations refer to dry gases ( 713 mmHg ), constant outside temperature and air pressure. The diffusion constants are shown in the Table 6. Diffusion surface: A $=15 \mathrm{~cm}$, middle ear volume: 9 ml . Diffusion processes are described by the Noyes-Whitney equations. Table 7. The ET performs pressure equalization every minute.

## Data and methods

The calculations refer to dry gases ( 713 Hgmm ), constant outside temperature and air pressure.

$$
\begin{aligned}
& \text { Diffusion constant: } D_{O_{2}}=20 \frac{\mathrm{~cm}}{\mathrm{~min} \cdot \mathrm{Hgmm}} \\
& D_{N_{2}}=4 \frac{\mathrm{~cm}}{\mathrm{~min} \cdot \mathrm{Hgmm}}
\end{aligned}
$$

Diffusion surface: $\mathrm{A}=15 \mathrm{~cm}^{2}$, middle ear volume: 9 ml .
The diffusion process is described by the Noyes-Whitney equations.

## Table 6.

## The equations

$$
\begin{aligned}
& \frac{d P_{O_{2}}(t)}{d t}=D_{O_{2}} \cdot A \cdot\left(P_{O_{2}}(t)-P_{O_{2} \text { tisulue }}\right) \\
& \frac{d P_{N_{2}}(t)}{d t}=D_{N_{2}} \cdot A \cdot\left(P_{N_{2}}(t)-P_{N_{\text {tissue }}}\right)
\end{aligned}
$$

Where the temporal change in the oxigen partial pressure $P_{o_{1}}(t)$, the temporal change in the nitrogen partial pressure $P_{N_{2}}(t)$ and the partial pressure of dissolved gases in tissues $P_{o, i s s s e}$ and $^{\text {and }} P_{\text {s, }}$ hisure

## Table 7.

Basic physical concepts: The gases flow across the ET according to their percentage composition from the higher pressure point to the lower pressure point.

The gases diffuse across the mucosa according to their partial pressure difference.
The process was analyzed for $\mathrm{O}_{2}$ and $\mathrm{N}_{2}$ gases only.

## RESULTS

The diffusion of gases at constant atmospheric pressure shows Table 8.

Gas exchange according to the classical theory (Hgmm)


## Table 8.

This is actually the classic theory. The middle ear is filled with air / 21\% oxygen, $79 \%$ nitrogen /. Oxygen diffuses in the ST according to its pressure gradient, the pressure in the ME cavity decreases. The ET opens and compensates for the pressure decrease with air ( $21 \%$ oxygen, $79 \%$
nitrogen). Therefore, the partial pressure of oxygen decreases and that of nitrogen increases in the ME (Figure 1,2.) As a result, at one point, more and more nitrogen diffuses into the surrounding tissues (Figure 3).


Figure 1.


Figure 2.


Figure 3.

If nitrogen is replaced by air, the process is reversed. The partial pressure of oxygen increases and that of nitrogen decreases. The two processes take place in parallel, then process stabilizes when the ratio of nitrogen to oxygen absorption is $21: 79$ as it is replaced or the same amount of gas is absorbed from oxygen and nitrogen. /Supporting information/ The partial
pressure changes of $\mathrm{O}_{2}$ and $\mathrm{N}_{2}$ and total pressure are shown figures 4-6. Gas exchange takes place in a similar way, starting from the optimal gas composition, only the equilibrium state is reached earlier in time. (Table 2,3).

## The time course of the change in total middle ear pressure based on the model



Figure 4.

Change in the partial pressure of nitrogen when opening the Eustachian tube (first 540 sec )


Figure 5.

# Change in the partial pressure of oxygen when the Eustachian tube is opened (first 540 sec ) 



## Figure 6.

A similar process takes place with increasing air pressure. When the ET opens, air flows into the middle ear. The partial pressures of $\mathrm{O}_{2}$ and $\mathrm{N}_{2}$ increase by 21:79, the absorption of $\mathrm{O}_{2}$ and $\mathrm{N}_{2}$ increases, absorption ratio changes slightly and then returns to 21:79. With constant and increasing air pressure, the partial pressure of carbon dioxide is essentially unchanged, which corresponds to the surrounding tissues.
The steady state lasts until the air pressure changes.
As the air pressure decreases, if the ET opens the gases flow outwardly through the ET according to their percentage composition and reduce the partial pressures of oxygen, carbon dioxide, and nitrogen in the ME. In the normal state, the partial pressure difference of $\mathrm{O}_{2}$ and $\mathrm{N}_{2}$ between the middle ear and the surrounding tissues is around $50-60 \mathrm{mmHg}$ /Supporting information/. This pressure difference is sufficient to compensate for the athmospheric pressure decrease. As the diffusion coefficient of carbon dioxide is the best ( 20 times better than oxygen /, a slight, transient pressure increase may occur in the middle ear.
In the event of a sudden drop in air pressure greater than $50-60 \mathrm{mmHg}$, if the partial pressures of $\mathrm{O}_{2}, \mathrm{~N}_{2}$, and $\mathrm{CO}_{2}$ in the ME fall below the level of the surrounding tissues, $\mathrm{O}_{2}, \mathrm{~N}_{2}$, and $\mathrm{CO}_{2}$ may diffuse into the middle ear. This can cause a transient, two-peak pressure rise in the ME. / $\mathrm{CO}_{2}$ faster and bigger wave, $\mathrm{O}_{2}$ and $\mathrm{N}_{2}$ slow, elongated curve./
Compensating for a sudden drop in air pressure is the most difficult task in middle ear ventilation.
With increasing and constant air pressure, the Eustachian tube equalizes towards the middle ear cavity, with decreasing air pressure, it equalizes towards the outside world, in which case the direction of diffusion may change according to the partial pressures differences in the middle ear. Because diffusion and flow through the ET is a passive process, the ME ventilation is controlled by the active ET opening.

The ET creates, the gas diffusion equalizes for the partial pressure differences between the middle ear and the surrounding tissues.

## CONCLUSIONS

## THE MIDDLE EAR VENTILATION CHARACTERISTICS.

Gas flow across the ET and gas diffusion between ME and ST is a twoway process.
The ME pressure decrease is always equalized with air $\left(21 \% \mathrm{O}_{2}, 79 \% \mathrm{~N}_{2}\right)$ The pressure decrease is caused by the absorption of $\mathrm{O}_{2}$ and $\mathrm{N}_{2}$ in the middle ear.
A pressure equalizes /air flow/ and the gas diffusion passive process
The gas diffusion continuous, the pressure equalization intermittent.
Diffusion of a certain amount of gas $\left(\mathrm{O}_{2}, \mathrm{~N}_{2}, \mathrm{CO}_{2}\right)$ causes a greater pressure decrease in a smaller volume in the ME.
In the ME the pressure decrease due to nitrogen or oxygen absorption is always compensated by air $/ 21 \%$ Oxygen, $79 \%$ Nitrogen/. When the absorbed oxygen or nitrogen is replaced by air, it changes the partial pressure of oxygen and nitrogen in the ME in the opposite direction.
Ventilation of the ME is the result of a complex, dynamic process controlled by the active ET opening. The ME is a closed cavity, which is opened time to time. This is the most important statement of ME ventilation.
We designed a valve tube that is closed and opens intermittently. Accordingly, we designed a tympanostomy tube that performs this task.

## VALVULAR TYMPANOSTOMY TUBE

The following figures 7-10 show the structure of valve tube

## The structure of the valve tube $I$.



Figure 7.

## The structure of the valve tube II.



Figure 8.

## Composition of the valve tube



Figure 9.

## Constructed valve tube



## Figure 10.

The temperature of the human body can vary widely. It is known that in muscle work, body temperature can be as high as 40 degrees Celsius. The valve container should be filled with a fluid boiling point around body temperature to move the valve diaphragm. The membrane material is flexible or flexible plastic. During physical activity, the body temperature
rises and the liquid enclosed in the tank boils, expands, which closes the valve. At rest, the body's temperature decreases, the fluid cools, its volume decreases, thus opening the valve. Diethyl ether ( C 4 H 10 O ) has a boiling point of $36.4^{\circ} \mathrm{C}$ and may be suitable for valve operation (Figure 11).

# The fluid, which is suitable for product the valve is e.g. ether data 



## Figure 11.

## DISCUSSION

Our valve tube is regulated by the volume change due to the change in body temperature, its preparation is a technical task.

The operation of the valve can be controlled by the full pressure difference, the partial pressure differences, the pulses of the chemomechanoreceptors. By setting the opening time, changes due to thesurface / volume ratio can also be compensated and monitored. The idea can probably be solved with nanotechnology.

## Abbreviations

ET :Eustachian tube
ME :middle ear
ST :surrounding tissues

PP :partial pressure
PPD :partial pressure difference

## Competing Interests

The authors declare that they have no competing interests.

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## Supporting Information

Gas exchange
At constant air pressure
Equilibrium can theoretically be achieved in two forms

## Steady state 1

Partial pressure difference between $\mathrm{O}_{2}$ and $\mathrm{N}_{2}$ in the middle ear.
The partial pressures of oxygen and nitrogen changes during pressure equalization until the same amount of gas is absorbed from oxygen as from nitrogen.
This is when the balance is reached
E.g.: $10-10 \mathrm{ml}_{2}$ and $\mathrm{N}_{2}$ is absorbed, and is replaced it with air

| $\mathrm{O}_{2} 21 \%$ |  | $\mathrm{O}_{2} \downarrow 2,1 \mathrm{ml}$ |  |
| :--- | :--- | :---: | :--- |
|  | $\Rightarrow \mathbf{O}_{2}$ | $\mathrm{~N}_{2} \uparrow 7,8 \mathrm{ml}$ |  |
| $\mathrm{N}_{2} 79 \%$ |  |  |  |
|  |  |  |  |
| $\mathrm{O}_{2} 21 \%$ |  | $\mathrm{O}_{2} \uparrow 2,1 \mathrm{ml}$ |  |
|  |  | $\mathbf{N}_{2}$ | $\mathrm{~N}_{2} \downarrow 7,8 \mathrm{ml}$ |
| $\mathrm{N}_{2} 79 \%$ |  | O |  |
|  |  | 2. |  |

So if I replace 10 ml of absorbed oxygen with air, the partial pressure of oxygen decreases, the nitrogen increases
If I replace 10 ml of absorbed nitrogen with air, the partial pressure of oxygen increases, the nitrogen decreases. The result is zero, steady state was created.
The amount of gas absorbed is given by the product of the partial pressure differences and the diffusion coefficients.
The same amount of gas is absorbed from $\mathrm{O}_{2}$ and $\mathrm{N}_{2}$ when the ratio of the partial pressure between $\mathrm{O}_{2}$ and $\mathrm{N}_{2}$ is 1:5, i.e. 5 times. In the middle ear, the partial pressure and the total pressure do not change. The result of the gas exchange is 0 i.e. zero
For example

| OXIGEN $/ \mathrm{mmHg} /$ | Absorbed volume of $\mathrm{O}_{2}$ | NITROGEN $/ \mathrm{mmHg} /$ | Absorbed volume of $\mathrm{N}_{2}$ |
| :--- | :--- | :--- | :--- |
| 1 PPDx 20 DC $=$ | 20 ml | 5 PPDx 4 DC $=$ | 20 ml |
| 10 PPD $\times 20 \mathrm{DC}=$ | 200 ml | 50 PPD $\times 4 \mathrm{DC}=$ | 200 ml |
| 20 PPD $\times 20 \mathrm{DC}=$ | $400 \mathrm{ml} \quad ? ?$ | 100 PPD $\times 4 \mathrm{DC}=$ | $400 \mathrm{ml} \quad ? ?$ |

The last line is incorrect. There are no such partial pressure differences between the middle ear and the surrounding tissues.
The above equilibrium is reached when the partial pressure difference of oxygen is between $1-10 \mathrm{mmHg}$ and that of nitrogen is between $5-50 \mathrm{mmHg}$ between the middle ear and the surrounding tissues.
Partial pressure in the middle: $\mathrm{O}_{2}=41-50 \mathrm{mmHg}, \mathrm{N}_{2}=578-623 \mathrm{mmHg}$
$\mathrm{O}_{2}: \mathrm{N}_{2}=1: 1$ absorption
$\mathrm{O}_{2}: \mathrm{N}_{2}=1: 5$ partial pressure difference
Diffusion constant : $\mathrm{O}_{2}: \mathrm{N}_{2}=20: 4=5$
PPD= partial pressure difference
Dc=diffusion constant
$\downarrow=$ partial pressure decrease
$\uparrow=$ partial pressure increase

## Steady state 2

As much is absorbed as it is replaced
Partial pressure differences between middle ear and surrounding tissues
If the partial pressure difference between the middle ear and the surrounding tissues for $\mathrm{O}_{2}$ and $\mathrm{N}_{2}$ gases is 1:20, then the absorption of $\mathrm{O}_{2}$ and $\mathrm{N}_{2}$ takes place in a ratio of 21:79 as it is replaced by air / 21: 79 / $\approx 4 \mathrm{x}$
For example
Partial pressure differences of $\mathrm{O}_{2}$ and $\mathrm{N}_{2}$ between ME and ST

| OXIGEN | Absorbed volume of $\mathbf{O}_{2}$ | NITROGEN | Absorbed volume of N $2 \mathbf{2}$ |
| :---: | :---: | :---: | :---: |
| $1 \mathrm{Hgmm} \times 20 \mathrm{DC}=$ | 20 ml | $20 \mathrm{Hgmm} \times 4 \mathrm{DC}=$ | 80 ml |
| $2 \mathrm{Hgmm} \times 20 \mathrm{DC}=$ | 40 ml | $40 \mathrm{Hgmm} \times 4 \mathrm{DC}=$ | 160 ml |
| $2,5 \mathrm{Hgmm} \times 20 \mathrm{DC}=$ | 50 ml | $50 \mathrm{Hgmm} \times 4 \mathrm{DC}=$ | 200 ml |
| $3 \mathrm{Hgmm} \times 20 \mathrm{DC}=$ | $60 \mathrm{ml} ? ?$ | $60 \mathrm{Hgmm} \times 4 \mathrm{DC}=$ | 240 ml ?? |

The last line is incorrect. The maximum partial pressure difference for $\mathrm{N}_{2}$ can be around 50 mmHg . This is the same as the measured and computer results measured. / 616 mmHg in the middle ear /
The absorption ratio of $\mathrm{O}_{2}$ and $\mathrm{N}_{2}$ is 21:79. / about 4 x / As much is absorbed as it is replaced.
The steady state lasts until the air pressure changes.
$\mathrm{O}_{2}: \mathrm{N}_{2}=21: 79 \approx 1: 4$ absorption
$\mathrm{O}_{2}: \mathrm{N}_{2}=1: 19 \approx 20$ partial pressure difference between ME and ST
This requires partial pressures of $41-42,5 \mathrm{mmHg}, \mathrm{O}_{2}$ and $593-623 \mathrm{mmHg} \mathrm{N}_{2}$ in the middle ear.
Steady state 1 equilibrium occurs with decreasing air pressure, steady state 2 with constant and increase air pressure.


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